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hybrid orbitals - ucla - of a hydrogen atom on methane. they may also contain lone pairs—this explains the geometry of water, which is sp^3 hybridized due to the lone pair, which occupies one of the four hybrid orbitals.
a guide to the anomeric carbon susan chang - ucla - what is an anomeric carbon? often times, a carbohydrate exists in a cyclic or an acyclic form. during cyclization, the carbonyl carbon transforms into a new stereocenter.

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